

Controlling the Selectivity in Ethenolysis of Cyclic Olefins with Ru-Based Metathesis Catalysts via Ancillary NHC Ligands: A Structure Activity Study

Pascal S. Engl, Celine B. Santiago, Christopher P. Gordon, Wei-Chih Liao, Alexey Federov, Christophe Copéret, Matthew S. Sigman, Antonio Togni

Departement of Chemistry and Applied Biosciences, ETH Zürich, Vladimir-Prelog-Weg 2, CH-8093 Zürich, Switzerland

Cross-metathesis reaction between cyclic olefins and ethylene (ethenolysis) yields valuable terminal dienes.¹ While enetholysis of ethyl oleate with Grubbs-type Ru metathesis catalysts has been widely studied² there currently are only a few reports on enetholysis of cyclic olefins with Ru catalysts,³ presumably due to the high activity of Grubbs-type initiators in the ring opening metathesis polymerization. Supporting ligands can greatly influence the selectivity of Ru catalysts.⁴ For instance, ruthenium metathesis catalysts with unsymmetrical N-trifluoromethyl NHC ligands display an unusually high selectivity (80–90%) for terminal olefins in enetholysis of ethyl oleate.⁵ In the present work we tested 28 isostructural Ru metathesis catalysts featuring various NHC ligands in enetholysis of *cis*-cyclooctene as a model substrate (Figure 1).

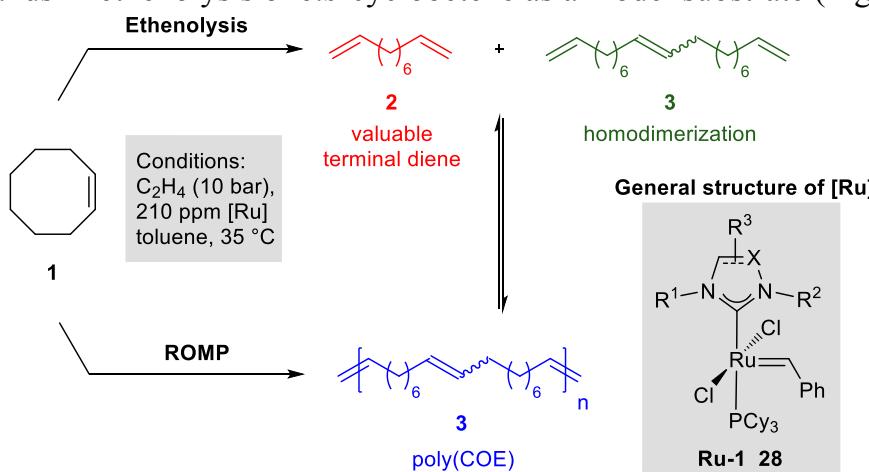


Figure 1. Ethenolysis of *cis*-cyclooctene. R¹, R², R³ = aryl, heteroaryl, cycloalkyl, alkyl, X=C, N.

Our study identified the dissymmetry of the NHC ligand as a key structural feature for improving the chemoselectivity to the terminal diene from 12% for the benchmark second-generation Grubbs catalyst up to 53% for the catalyst containing an unconventional CF₃-functionalized NHC ligand. To rationalize the experimental results, a computational study was conducted that explained the observed dependence via the univariate correlations of activity to electronic and steric properties of the ligand. The optimal catalyst compared favorably to the Grubbs 2nd generation catalyst in enetholysis of other cyclic olefins including functionalized norbornenes.

- [1] Marinescu, S. C.; Schrock, R. R.; Müller, P.; Hoveyda, A. H., *J. Am. Chem. Soc.* **2009**, *131*, 10840 - 10841.
- [2] (a) Bidange, J.; Fischmeister, C.; Bruneau, C., *Chem. Eur. J.* **2016**, *22*, 12226 - 12244; (b) Thomas, R. M.; Keitz, B. K.; Champagne, T. M.; Grubbs, R. H., *J. Am. Chem. Soc.* **2011**, *133*, 7490 - 7496.
- [3] (a) Kiss, L.; Kardos, M.; Forró, E.; Fülöp, F., *Eur. J. Org. Chem.* **2015**, 1283 - 1289 ; (b) Eyrisch, K. K.; Müller, B. K. M.; Herzig, C.; Nuyken, O., *Designed Monomers and Polymers* **2004**, *7*, 661 - 676; (c) Eyrisch, K. K.; Müller, B. K. M.; Herzig, C.; Nuyken, O., *J. Organometallic Chem.* **2000**, *606*, 3 - 7
- [4] Vougioukalakis, G. C.; Grubbs, R. H., *Chem. Rev.* **2010**, *110*, 1746 - 1787.
- [5] Engl, P. S.; Fedorov, A.; Copéret, C.; Togni, A., *Organometallics* **2016**, *35*, 887 - 893.
- [6] Engl, P. S.; Santiago, C.; Gordon, C. P.; Wei-Chih, L.; Fedorov, A.; Sigman, M. S.; Copéret, C.; Togni, A., *J. Am. Chem. Soc.* **2017**, *139*, 13117 - 12125.